

(1.1)

Free body diagram

$$F = G \frac{mM}{R^2} = \left(\frac{GM}{R^2} \right) m$$

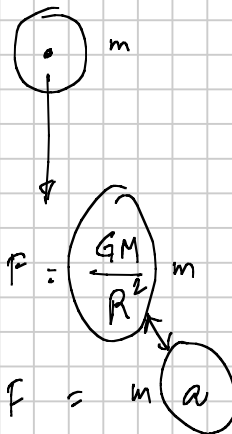
$$F = \frac{mg}{\text{with } \frac{m}{s^2}} [N]$$

\downarrow 9.81

$$F = \frac{mg}{1000} [kN]$$

$$F = \left(\frac{GM}{R^2} \right) m$$

\downarrow g



Therefore, $\frac{GM}{R^2}$ must be the accel of free fall.

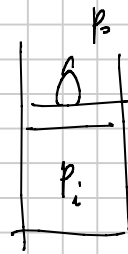
$$g = \frac{GM}{R^2} = 9.81 \frac{m}{s^2}$$

Benchmark for force : 1 kN is the weight of a 100 kg person on earth.

$$p_0 = 101 \text{ kPa} \approx 100 \text{ kPa} \left[\frac{kN}{m^2} \right]$$

$$P_i A = P_o A + \frac{mg}{1000}$$

$\underbrace{\text{KN} \cdot \text{m}^2}_{= \text{KN}} \quad \underbrace{\text{KN}} \quad \underbrace{\text{KN}}$



$$P_i = P_o + \frac{mg}{1000 A}$$

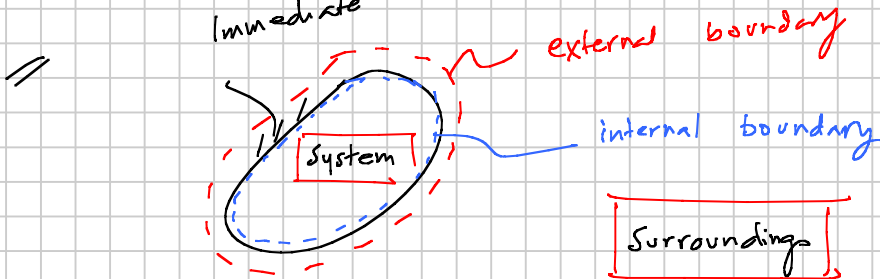
$\underbrace{100 \text{ kPa}} \quad \underbrace{\text{kg}} \quad \underbrace{10 \frac{\text{m}}{\text{s}^2}} \quad \underbrace{\text{m}^2}$

// (1.2)

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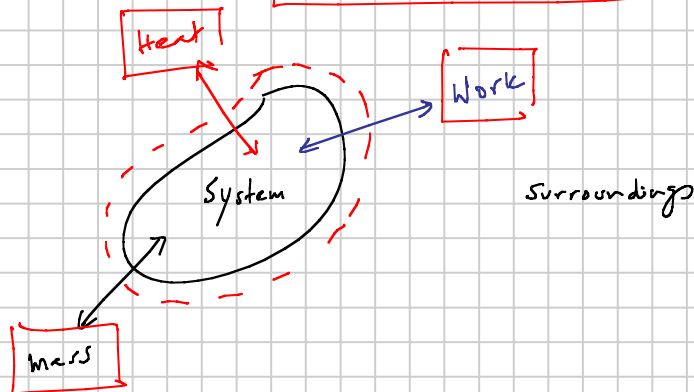


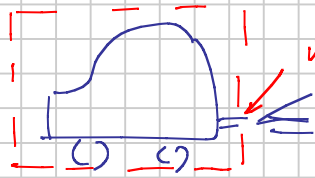
Immediate surroundings



System + Immediate Surroundings

= System's Universe.

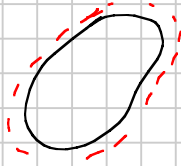




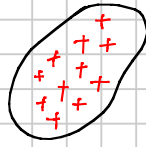
What interaction is this?

Obvious answer should be Mars

Mars Transfer :



Mars m [kg]



Mole n [kmol]

A pure Substance means composition

does not change from place to place.

$$\frac{m}{n} \left[\frac{\text{kg}}{\text{kmol}} \right] \text{ must}$$

be a constant for a pure substance.

A simple system is made of a pure substance.

Molar mass is mass in kg per Dozen aka kmol of a pure substance.

$$\bar{M} = \frac{m}{n} \left[\frac{\text{kg}}{\text{kmol}} \right]$$

$$\Rightarrow n = \frac{m}{\bar{M}}$$

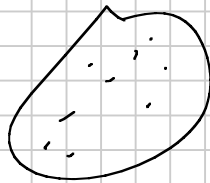
$$\bar{M}_{H_2} = 2 \frac{\text{kg}}{\text{kmol}}$$

$$\bar{M}_{O_2} = 32 \frac{\text{kg}}{\text{kmol}}$$

$$\bar{M}_{N_2} = 28 \frac{\text{kg}}{\text{kmol}}$$

$$\bar{M}_{CO_2} = 44 \frac{\text{kg}}{\text{kmol}}$$

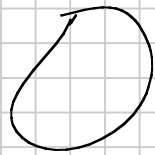
$$\bar{M}_{Ar} = 39 \frac{\text{kg}}{\text{kmol}}$$



$$\frac{m}{V} \left[\frac{\text{kg}}{\text{m}^3} \right] \rho = \frac{m}{V} \left[\frac{\text{kg}}{\text{m}^3} \right]$$

$$\underbrace{V}_{\text{specific volume}} = \frac{V}{m} \left[\frac{\text{m}^3}{\text{kg}} \right] = \frac{1}{\rho}$$

m, n, p, v, \bar{M}



Volume = $V \text{ m}^3$
 mole $n \text{ kmol}$
 mass $m \text{ kg}$

$$v = \frac{V}{m} \left[\frac{\text{m}^3}{\text{kg}} \right]$$

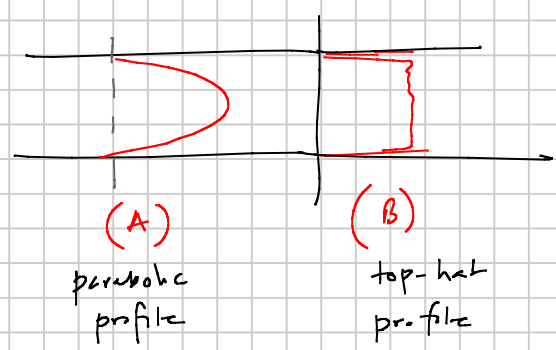
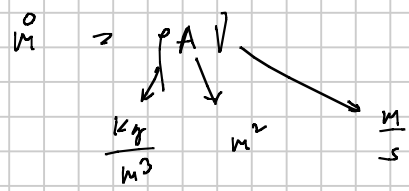
$$\bar{v} = \frac{V}{n} \left[\frac{\text{m}^3}{\text{kmol}} \right]$$

$$\bar{v} = \frac{V}{n} = \frac{V}{(m/\bar{M})}$$

$$= \frac{V \bar{M}}{m} = \bar{M} v$$

$\therefore \boxed{\bar{v} = \bar{M} v}$ [Advanced concept]

== Pipes & tubes & ducts carry mass



== $\dot{m} = \rho A V$ [benchmark]

